QUARTERLY STATUS REPORT

ON

EFFICACY OF ALKALI - SUPEROXIDE BEDS

FOR BACTERIA REMOVAL FROM AIR

CONTRACT R-21-009-007

March 1, 1966 - May 31, 1966

Prepared by:	Stanley J. Burdick	Date June 17,1966
Approved by:	Dr. R. C. Evans	Date June 17, 1966
Approved by:	Dr. R. G. Bartlett	Date Jun 21, 1966
•	N66 85059 (ACCESSION NUMBER)	(CODE)
	THE CR OR TMX OR AD NUMBER)	(CATEGORY)

EFFICACY OF ALKALI - SUPEROXIDE BEDS FOR BACTERIA REMOVAL FROM AIR

I. INTRODUCTION

The potassium superoxide, then-bed canister has been subjected to a second series of systematic tests. This effort was undertaken to obtain additional data on the trends indicated by the variance of the four parameters in the first series of experimental tests. Figure 1 shows the complete factorial experiment for all possible interactions. The 13 tests indicated by an X in Figure 1 show the recently completed set of experiments.

II. SUMMARY OF EXPERIMENTAL RESULTS

The data collected and calculated from the second series of experiments are summarized in Table I. The first six columns give the conditions under which the experiment was performed. The remaining columns are calculated values indicating the overall performance of the canister.

The general trends resulting from a preliminary evaluation of the first test series appear to be confirmed by the data from the second test series. Increasing oxygen evolution and carbon dioxide absorption result from increasing relative humidity. The thinner beds lead to more efficient oxygen evolution and carbon dioxide absorption.

Increasing the rate of gas flow results in increasing oxygen evolution and decreasing carbon dioxide absorption. Oxygen evolution increases with decreasing particle size but the effect of particle size on carbon dioxide absorption is still inconclusive. A more detailed evaluation of all of the data will be given in the final report.

III. FUTURE PLANS

Supplementary experiments testing the efficacy of the superoxide canister toward bacteria are planned. Previous data indicated that the physical effect of the superoxide toward bacteria was greater than the chemical effect. A further elucidation of the mechanism by which the potassium superoxide destroys the bacteria was considered desirable.

The Johns Hopkins University

APPLIED PHYSICS LABORATORY

Silver Spring, Maryland

% RELATIVE		40%			52%			70%		97%			
BED DEPTH	Ges:Flow (cc/mir	250	565	1000	250	565	1000	250	565	1000	250	565	1000
	7-12	х											
1/8 Inch	12-24												
THEN	24-42			x									
	42-80				х								
	80-170					x							
1/4 Inch	7-12						x						
	12-24							Х					
	24-42												
	42-80					`				Х			
	80-170										х		
	7-12											х	
1/2 Inch	12-24										<u> </u>		х
	24-42											х	
	42-80			<u> </u>						1	X		
	80-170									х			

FIGURE 1
COMPLETION OF SYSTEMATIC SELECTION
OF EXPERIMENTS

TABLE I EXPERIMENTAL RESULTS

Overall R. Q.	ı	.49	.10	.21	.18	.75	1.01	.72	. 73 .	3 -	*84	.85	. 54	
Run Iime Iime to 5% (min)CO ₂ (min)	>280	>280	>280	>280	>280	>280	234	>280	198	151	>280	74	>280	
Run Time (min)	280	280	280	280	280	280	280	280	280	280	280	280	280	
% 0 Run Evolved Time (min)	12,8	19.2	8.64	25.6	24.2	43.9	89.1	73.4	61,1	82,5	21,5	81.9	23.4	
Evolved 0 (Titers)	1,333	4.432	10,859	5,235	4.374	3,602	7,756	11,349	9,228	4.227	1,274	4,331	1,913	
Absorbed CO ₂ (1fters)	•	2,155	1,087	1,086	.813	2,714	7,890	8,218	6,714	3,361	1,074	3,702	1,037	
Weight KO ₂ (gms.)	57,10	117,40	103,00	98,00	91,80	40.80	47.80	79.80	72.90	24.40	28,90	25,10	39,30	
Initial % CO ₂	4.07	3.97	4.22	3,94	4.05	3,85	3,94	3,75	4.15	3.84	3,94	3.90	3.81	
Mesh Initia (Tyler % CO ₂ Scale)	12-24	7-12	12-24	24-42	42-80	80-170	42-80	80-170	7-12	80-170	42-80	24-42	7-12	
Bed Depth (inch)	1/4	1/2	1/2	1/2	1/2	1/4	1/4	1/2	1/4	1/8	1/8	1/8	1/8	
Gas Flow (cc/min)	250	565	1000	565	250	250	1000	1000	1000	565	250	1000	250	
% R.H.	70.5	91.5	0.86	97.5	5.96	0.96	71.5	70.0	54.0	52.0	52.5	42.5	43.0	
Run #	116	117	118	119	120	121	122	123	124	125	126	127	128	